



Characterization of the Structure of *tert*-butyl[1-hydroxy-2-methyl-3-(1*H*-1,2,4-triazol-1-yl)]propan-2-ylcarbamate Using 2D Heteronuclear NMR Experiments

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Abstract

The alkylation of 1*H*-1,2,4-triazole with an *O*-tosyloxazoline derivative, followed by an oxazoline ring-opening reaction and protection of amine function, leads normally to obtain two regioisomers of β -aminoalcohols. After purification by column chromatography of the crude reaction product, only a single product is obtained. Hence, there is need of its identification by spectroscopic study.

Keywords

Oxazoline, 1*H*-1,2,4-triazole, β -aminoalcohol, 2D heteronuclear NMR experiments

Introduction

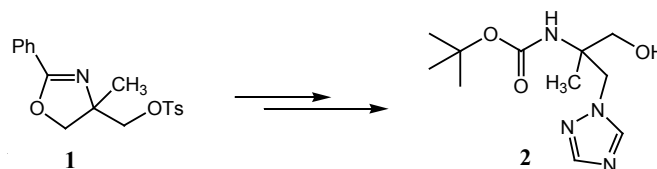
Triazoles constitute an important class of biologically active heterocyclic compounds that have received a great deal of attention since their discovery. Diverse compounds derived from 1,2,4-triazoles have a wide spectrum activities, including antimicrobial [1,2] and antibacterial properties [3,4], human antifungal agents [5], anticancer agents [6], antiviral [7], antitumor activity [8] and in agricultural science as potent fungicides, herbicides and insecticides [9,10]. Amino acids containing the 1,2,4-triazole moiety and their derivatives represent a well-known group of organic compounds also presenting biological activity. Thus β -(1,2,4-triazol-1-yl)-L-alanine is known as an important metabolite in plants of the fungicide myclobutanil [11-13] and β -(3-amino-1,2,4-triazol-1-yl)-L-alanine is a metabolite of the weed killer 3-amino-1,2,4-triazole [14].

Materials and Methods

NMR spectra (¹H, ¹³C and ¹⁵N) were recorded on a Bruker AM 300 (operating at 300.13 MHz for ¹H, at 75.47 MHz for ¹³C and at 30.41 MHz for ¹⁵N) spectrometer (Centre Universitaire Régional d'Interface, Fez-Morocco). NMR data are listed in ppm and are reported relative to tetra-methylsilane (¹H, ¹³C).

Discussion

It is reported that afforded the corresponding 1- and 4-alkylated isomers, with prevalence of the N₁-isomer [15-17]. Reaction of 1*H*-1,2,4-triazole with **1** and K₂CO₃, was carried out in the presence of a catalytic amount of tetrabutylammonium bromide in *N,N*-dimethylformamide at 120°C for 12 hours. The previous reaction stage is followed by an oxazoline ring-opening reaction carried out in acidic medium. The aminoalcohol derivative **2**, which is the aim of this paper, is obtained after addition of *tert*-butoxycarbonyl anhydride Boc to the intermediate product in a mixture of water/dioxane (1/2) at (0 < T < 5°C) in the presence of triethylamine [18]. Its structure was established on the basis of NMR spectroscopy (¹H, ¹³C and ¹⁵N), in addition to MS data and elemental analysis.



The definite assignment the chemical shifts of protons, carbons and nitrogens of compound **2** are shown in table 1 and table 2 (Figure 1,

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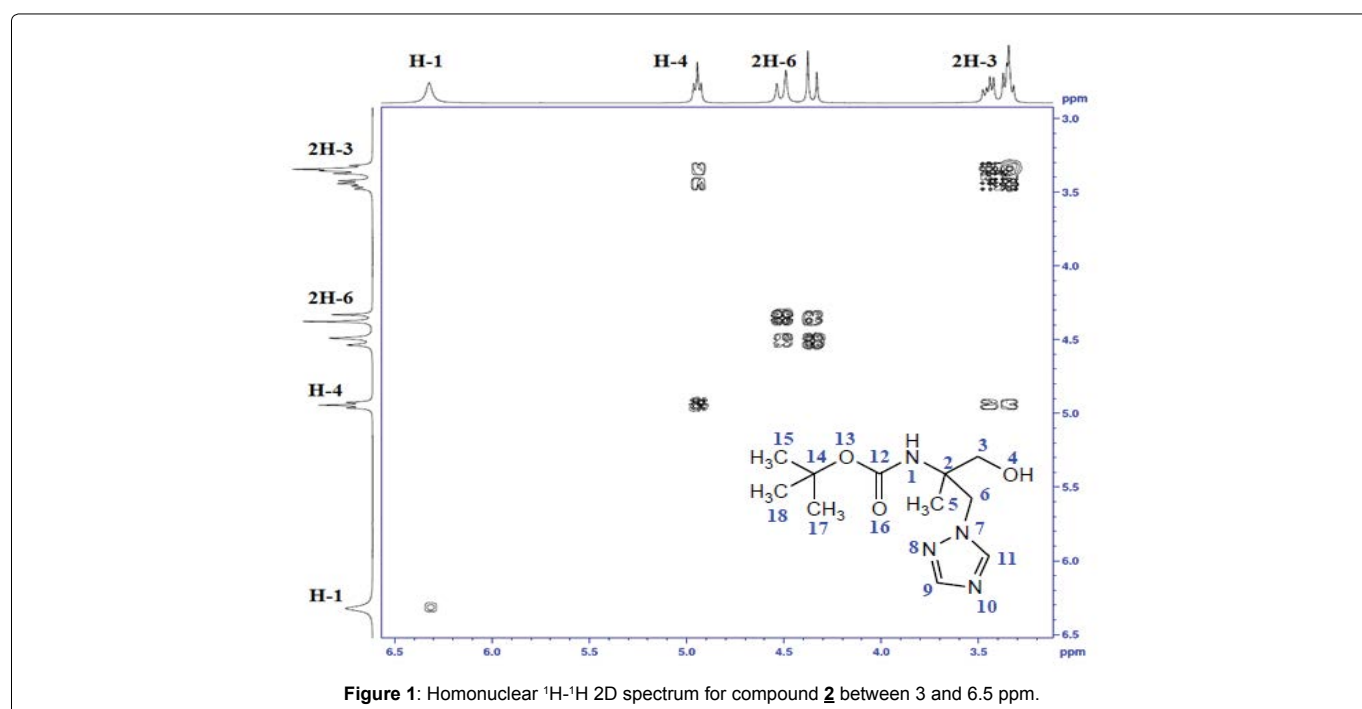
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Table 1: ^1H (300 MHz) and ^{13}C (75.47 MHz) NMR spectral data for **2** in DMSO-*d*₆, including results obtained by homonuclear 2D shift-correlated ^1H COSY and heteronuclear 2D shift-correlated HMQC ($^1J_{\text{CH}}$)^a. Chemical shifts (δ , ppm) and coupling constants (J , Hz, in parenthesis)^b.

Position	δ_{H}	δ_{C}	Correlation	
			^1H - ^1H	^1H - ^{13}C
(NH) 1	6.29 (s)	-	1-H	-
2	-	56.66	-	-
3	3.35, 3.45 (AB (2dd), 10.8; 5.6)	64.85	H ¹ -3, H ² -3, O-H	H ¹ -3, H ² -3, C-3
(OH) 4	4.92 (t, 5.6)	-	O-H, H ¹ -3, H ² -3	-
5	1.18 (s)	20.43	H ¹ -5, H ² -5, H ³ -5	H ^{1,2,3} -5; C-5
6	4.35, 4.51 (AB, 14)	51.74	H ¹ -6, H ² -6	H ¹ -6, H ² -6, C-6
9	7.95 (s)	151.61	H ³ triazole -9	H ³ triazole -9, C-9
11	8.23 (s)	145.24	H ⁵ triazole -11	H ⁵ triazole -11, C-11
12	-	154.97	-	-
14	-	78.43	-	-
15; 17; 18	1.39 (s)	28.68	H ^{1,2,3} -15	H ^{1,2,3} -15; C-15
			H ^{1,2,3} -17	H ^{1,2,3} -17; C-17
			H ^{1,2,3} -18	H ^{1,2,3} -18; C-18

^aCorrelation from C to the indicated hydrogens.^bChemical shifts and coupling constants (J) obtained of 1D ^1H -NMR spectrum.**Figure 1:** Homonuclear ^1H - ^1H 2D spectrum for compound **2** between 3 and 6.5 ppm.**Table 2:** Listing of ^{15}N (400 MHz) NMR spectral data for **2** in DMSO-*d*₆, including results obtained by heteronuclear single quantum coherence shift-correlated (HSQC) and heteronuclear multiple bond coherence shift-correlated (HMBC).

Position	δ_{H}	δ_{N}	Correlation
(NH) 1	7.02 (s)	93.29	H-1, N-1
5	1.24 (s)	93.29	H ^{1,2,3} -5, N-1
6	4.39, 4.83 (AB, 14)	93.29	H ^{1,2} -6, N-1
		214.54	H ^{1,2} -6, N-7
		299.12	H ^{1,2} -6, N-8
9	7.96 (s)	214.54	H ³ triazole-9, N-7
		251.97	H ³ triazole-9, N-10
		299.12	H ³ triazole-9, N-8
11	8.20 (s)	214.54	H ⁵ triazole-11, N-7
		251.97	H ⁵ triazole-11, N-10

Chemical shifts (δ , ppm) and coupling constants (J , Hz, in parenthesis) obtained of 1D ^1H -NMR spectrum. ^{15}N NMR spectrum of the compound **2** shows four signals corresponding to four nitrogen atoms and assigning each signal to its corresponding nitrogen atom is carried out based mainly on the spectral data ^1H - ^{15}N HMBC (Figure 3).

Figure 2 and Figure 3).

This interaction ^1H - ^{15}N has allowed us to make the following observations:

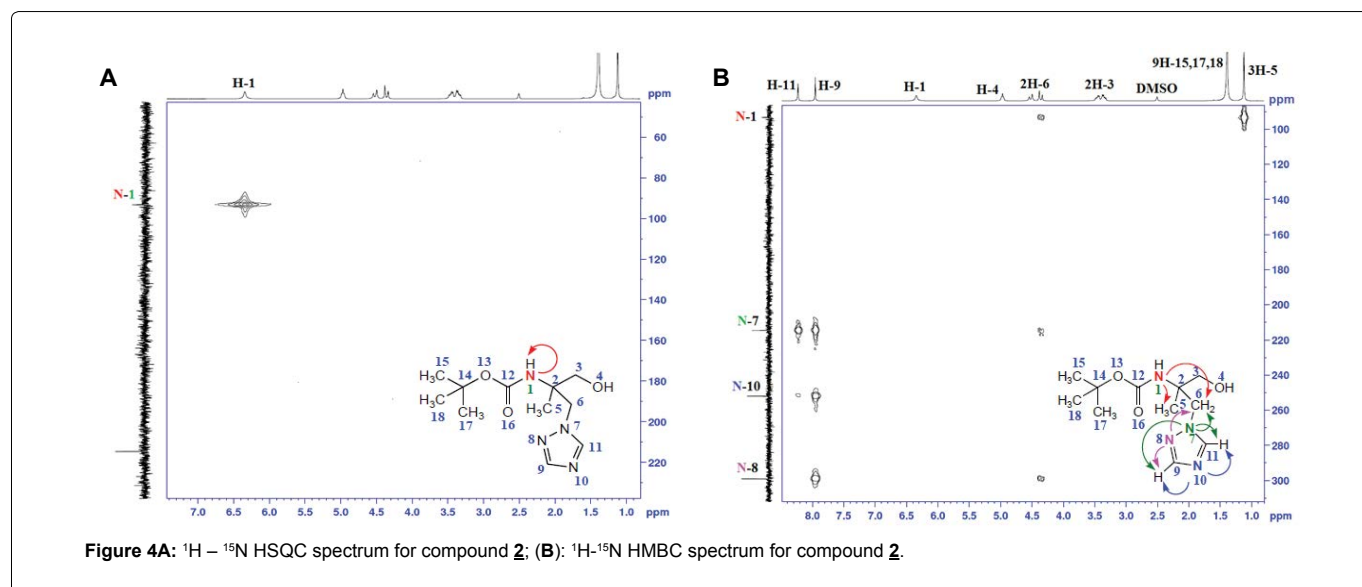
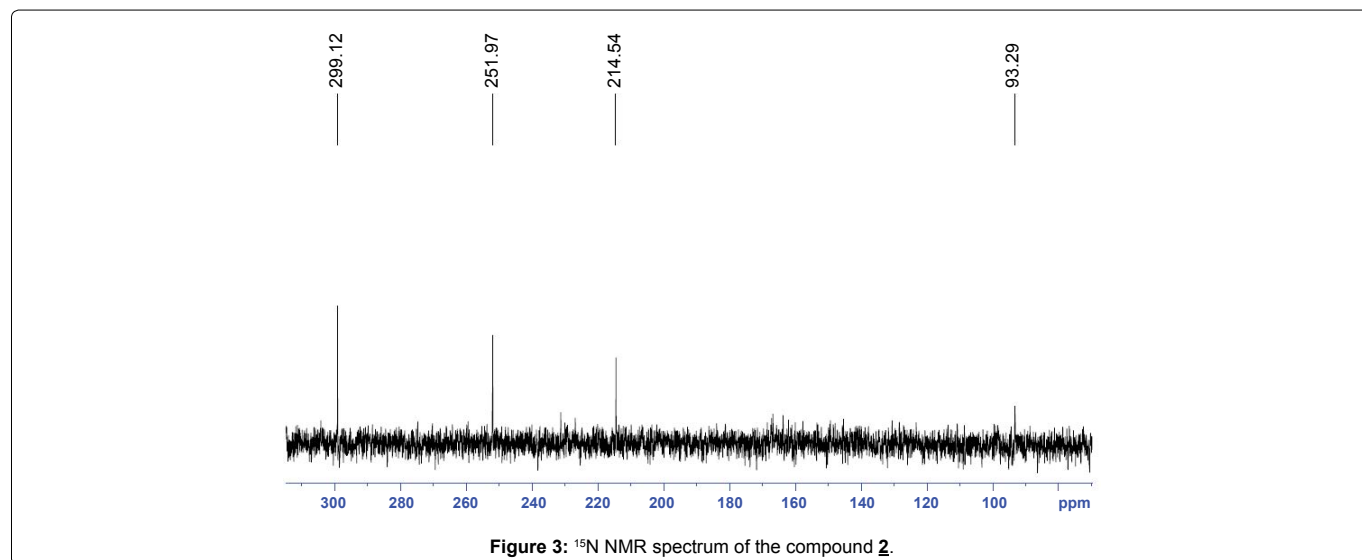
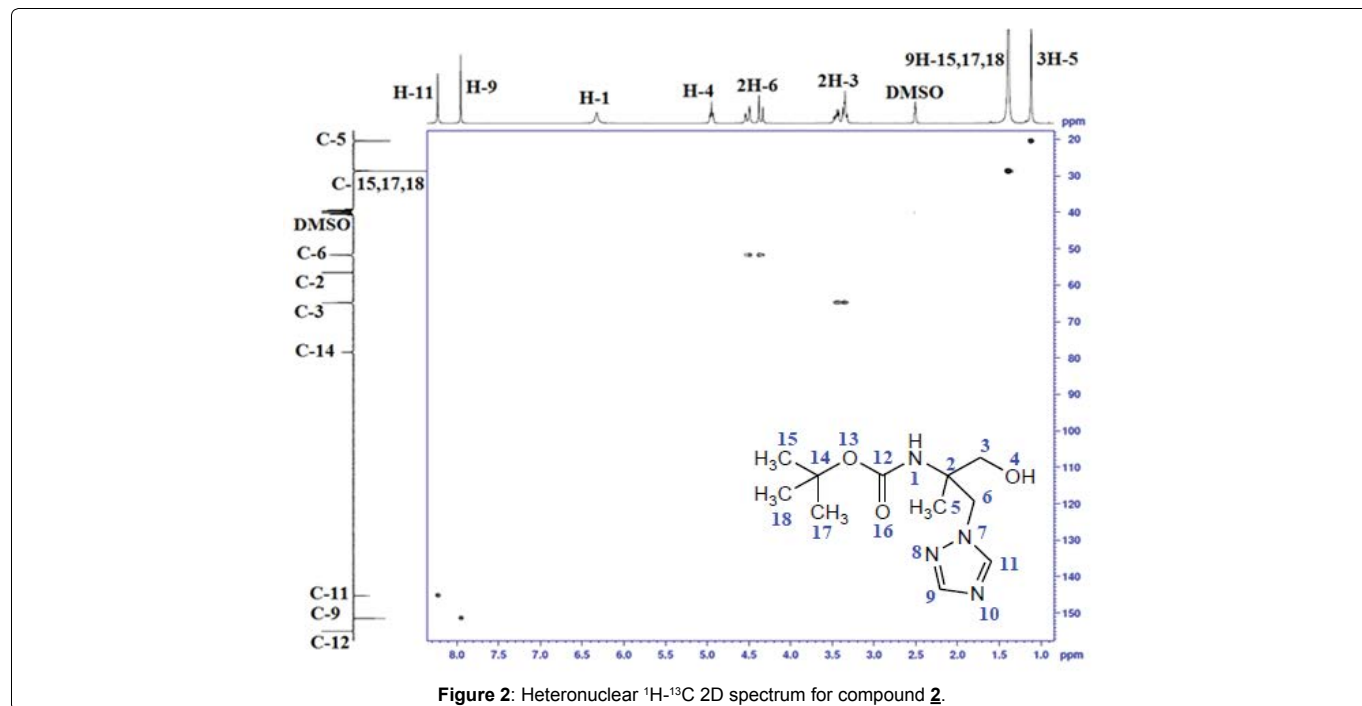
- A signal at 93.29 ppm attributed to the carbamate nitrogen N-1: correlation between CH_3 , CH_2 -triazole and the amidic nitrogen.

- A signal at 214.54 ppm attributed to the N-7 of the 1,2,4-triazole ring: correlation between the two triazole protons, CH_2 -triazole and N-7 nitrogen at position 1 of the 1,2,4-triazole ring.

- A signal at 251.97 ppm attributed to the N-10 of the 1,2,4-triazole ring: interaction between the two triazole protons and the N-10 located in the position 4 of the 1,2,4-triazole ring.

- A signal at 299.12 ppm attributed to N-8 of the 1,2,4-triazole ring: interaction between CH_2 -triazole, triazole proton H-9 and N-8 at position 2 of the 1,2,4-triazole ring.

Further, the analysis of ^1H - ^{15}N HMBC spectrum of compound **2** confirms that the nucleophilic substitution reaction of 1,2,4-triazole



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and *O*-tosyloxazoline derivative **1** is carried out on the nitrogen in position 1 of the 1,2,4-triazole ring and its structure has been clearly identified (Figure 4).

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